OZONOLYSIS

oxidative cleavage

$$\begin{array}{c} \text{CH}_{3} \\ \text{H} \end{array} \begin{array}{c} \text{CH}_{3} \\ \text{CH}_{3} \end{array} \begin{array}{c} \text{CH}_{3} \\ \text{O} \end{array} \begin{array}{c} \text{O} \\ \text{O} \end{array} \begin{array}{c} \text$$

An unknown compound (**A**) has a formula of C_7H_8 . Treatment of **A** with H_2 on Lindlar's catalyst gives compound **B** (C_7H_{12}). Treatment of **A** with H_2/Pd on carbon (standard hydrogenation) gives compound **C** (C_7H_{16}). Ozonolysis of **B** followed by a Zn/acetic work-up gives pentanedial and 2 equivalents of formaldehyde.

Propose structures for A, B, and C that are consistent with these data.